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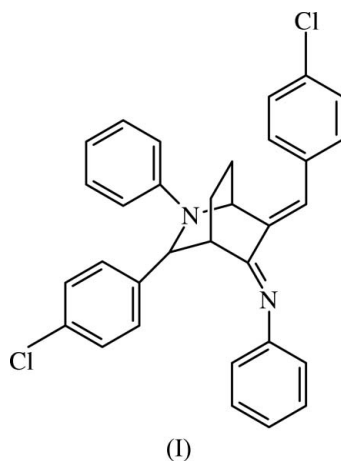
Key indicators

Single-crystal X-ray study
 $T = 273$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.055
 wR factor = 0.163
Data-to-parameter ratio = 14.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.***N*-{3-(4-Chlorophenyl)-6-[(*Z*)-(4-chlorophenyl)-methylidene]-2-phenyl-2-azabicyclo[2.2.2]oct-5-ylidene}aniline**The title compound, $\text{C}_{32}\text{H}_{26}\text{N}_2\text{Cl}_2$, exists as the *Z* isomer. All three rings in the azabicyclo[2.2.2]octane system adopt a boat conformation. Intermolecular $\text{C}-\text{H}\cdots\text{Cl}$ interactions form a dimer in the crystal structure.Received 12 December 2005
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Imino Diels–Alder adducts. X.

Comment

In continuation of our ongoing structural elucidation of azabicyclo[2.2.2]octane derivatives obtained from the Diels–Alder reaction, we report here the crystal structure of the title compound, (I).

The molecular geometry of (I) (Fig. 1), in terms of bond distances and angles, is in good agreement with that of similar structures (Sonar *et al.*, 2004; Ravikumar *et al.*, 2005*a,b*). The torsion angles $\text{C}5-\text{C}6-\text{C}27-\text{C}28 = -176.2$ (2)° and $\text{C}5-\text{C}6-\text{N}2-\text{C}21 = 176.4$ (2)° indicate a *trans* configuration, and hence the *Z* isomer. Due to the double-bond strain, the bond angles $\text{C}4-\text{C}5-\text{N}2$, $\text{C}6-\text{C}27-\text{C}28$ and $\text{C}1-\text{C}6-\text{C}27$ deviate from ideal values (Table 1).All three six-membered rings of the azabicyclo[2.2.2]octane system adopt the expected boat conformation, with asymmetry parameters (Nardelli, 1983) $\Delta C_s(\text{C}1) = 0.009$ (2), 0.017 (2), 0.027 (1) for the rings $\text{C}1/\text{C}4-\text{C}8$, $\text{N}1/\text{C}1/\text{C}3/\text{C}4/\text{C}7/\text{C}8$ and $\text{N}1/\text{C}1/\text{C}3-\text{C}6$, respectively. The $\text{C}9-\text{C}14$ phenyl ring and $\text{C}15-\text{C}20$ chlorophenyl ring make dihedral angles with the mean plane of atoms $\text{C}1/\text{C}3/\text{C}4/\text{N}1$ of the azabicyclo[2.2.2]octane cage of 30.7 (1) and 66.7 (1)°, respectively. Similarly, the dihedral angles are 77.9 (1) and 35.1 (1)°, respectively, for the $\text{C}21-\text{C}26$ phenyl ring and $\text{C}28-\text{C}33$ chlorophenyl ring with respect to the $\text{C}1/\text{C}4/\text{C}5/\text{C}6$ mean plane of the cage. The torsion angle $\text{C}9-\text{N}1-\text{C}3-\text{C}15$ is -90.2 (2)°, but $\text{N}2-\text{C}5-\text{C}6-\text{C}27$ is only 2.8 (4)°, as a consequence of the double bonds.

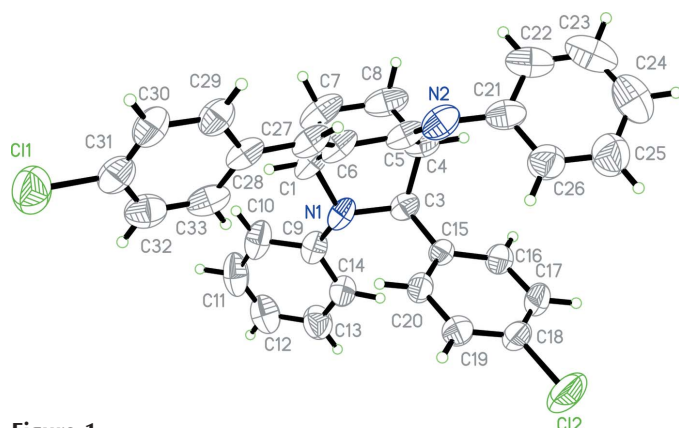


Figure 1

A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

The crystal structure is stabilized by weak intermolecular C—H···Cl interactions (You *et al.*, 2004) (Table 2). One of them leads to formation of a centrosymmetric dimer with a characteristic $R_2^2(26)$ motif (Bernstein *et al.*, 1995) in the crystal packing (Fig. 2).

Experimental

To a solution of *N*-benzylideneaniline (5.5 mmol) in CH_2Cl_2 (5 ml) at room temperature were sequentially added ZrCl_4 (10 mol%) and cyclohexen-2-one (5.5 mmol). The mixture was stirred for 6 h. After completion of the reaction as indicated by thin-layer chromatography, the reaction was quenched with water, and the mixture diluted with CH_2Cl_2 (20 ml) and washed with water (10 ml). The aqueous layer was extracted with CH_2Cl_2 (2×10 ml). The combined organic layers were dried over Na_2SO_4 and concentrated *in vacuo* to yield a dark-brown residue, which was purified by column chromatography using 2–10% ethyl acetate in hexane as eluent to obtain the pure product.

Crystal data

$\text{C}_{32}\text{H}_{26}\text{Cl}_2\text{N}_2$
 $M_r = 509.45$
 Monoclinic, $P2_1/n$
 $a = 14.3560$ (6) Å
 $b = 10.0621$ (4) Å
 $c = 18.3872$ (8) Å
 $\beta = 91.427$ (1)°
 $V = 2655.24$ (19) Å³
 $Z = 4$

$D_x = 1.274$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 5389 reflections
 $\theta = 2.2$ – 23.0 °
 $\mu = 0.27$ mm⁻¹
 $T = 273$ (2) K
 Block, colorless
 $0.20 \times 0.12 \times 0.07$ mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 ω scans
 Absorption correction: none
 18622 measured reflections
 4657 independent reflections

3312 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.021$
 $\theta_{\text{max}} = 25.0$ °
 $h = -17 \rightarrow 17$
 $k = -11 \rightarrow 11$
 $l = -21 \rightarrow 21$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.163$
 $S = 1.03$
 4657 reflections
 325 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0676P)^2 + 0.9266P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.37$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.36$ e Å⁻³

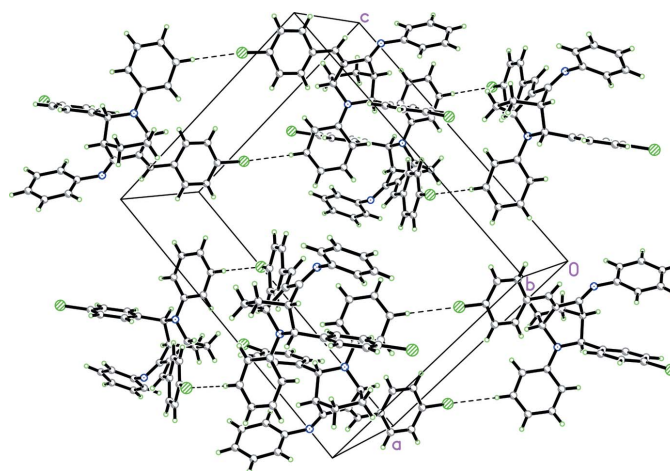


Figure 2

A view of the packing, showing the dimers. Dashed lines indicate C—H···Cl interactions.

Table 1

Selected geometric parameters (Å, °).

C11—C31	1.737 (4)	N2—C5	1.267 (4)
Cl2—C18	1.745 (3)	C6—C27	1.333 (4)
N2—C5—C4	128.3 (3)	C6—C27—C28	131.1 (3)
C27—C6—C1	128.5 (3)		

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C11—H11···Cl1 ⁱ	0.93	2.78	3.706 (3)	177
C10—H10···Cl2 ⁱⁱ	0.93	3.06	3.974 (3)	167

Symmetry codes: (i) $-x + 1, -y + 1, -z + 2$; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

H atoms were included in calculated positions ($C-H = 0.93$ – 0.98 Å) and refined as riding, with $U_{\text{iso}}(\text{H})$ values set at 1.2 or 1.5 (CH_3) times $U_{\text{eq}}(\text{C})$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL/PC* (Sheldrick, 1990); software used to prepare material for publication: *SHELXL97*.

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References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bruker (2001). *SAINT* (Version 6.28a) and *SMART* (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.
 Nardelli, M. (1983). *Acta Cryst.* **C39**, 1141–1142.
 Ravikumar, K., Sridhar, B., Mahesh, M. & Narayana Reddy, V. V. (2005a). *Acta Cryst.* **E61**, o2277–o2279.
 Ravikumar, K., Sridhar, B., Mahesh, M. & Narayana Reddy, V. V. (2005b). *Acta Cryst.* **E61**, o3273–o3275.

Sheldrick, G. M. (1990). *SHELXTL/PC* User's Manual. Bruker AXS Inc., Madison, Wisconsin, USA.

Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

Sonar, V. N., Parkin, S & Crooks, P. A. (2004). *Acta Cryst. C60*, o659–o661.

You, X. L., Lu, C. R., Zhang, Y. & Zhang, D. C. (2004). *Acta Cryst. C60*, o693–o695.